

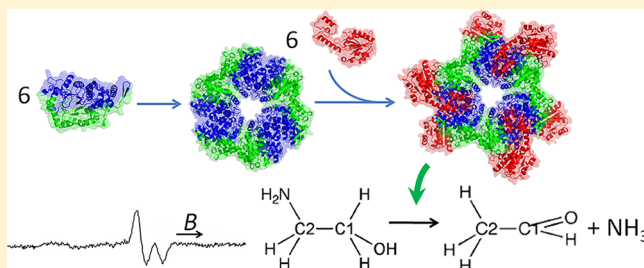
The Structural Model of *Salmonella typhimurium* Ethanolamine Ammonia-Lyase Directs a Rational Approach to the Assembly of the Functional [(EutB–EutC)₂]₃ Oligomer from Isolated Subunits

Adonis Miguel Bovell and Kurt Warncke*

Department of Physics, Emory University, Atlanta, Georgia 30322, United States

S Supporting Information

ABSTRACT: Ethanolamine ammonia-lyase (EAL) is a 5'-deoxyadenosylcobalamin-dependent bacterial enzyme that catalyzes the deamination of the short-chain vicinal amino alcohols, aminoethanol and (S)- and (R)-2-aminopropanol. The coding sequence for EAL is located within the 17-gene *eut* operon, which encodes the broad spectrum of proteins that comprise the ethanolamine utilization (*eut*) metabolosome suborganelle structure. A high-resolution structure of the ~500 kDa EAL [(EutB–EutC)₂]₃ oligomer from *Escherichia coli* has been determined by X-ray crystallography, but high-resolution spectroscopic determinations of reactant intermediate-state structures and detailed kinetic and thermodynamic studies of EAL have been conducted for the *Salmonella typhimurium* enzyme. Therefore, a statistically robust homology model for the *S. typhimurium* EAL is constructed from the *E. coli* structure. The model structure is used to describe the hierarchy of EutB and EutC subunit interactions that construct the native EAL oligomer and, specifically, to address the long-standing challenge of reconstitution of the functional oligomer from isolated, purified subunits. Model prediction that the (EutB)₂ oligomer assembly will occur from isolated EutB, and that this hexameric structure will template the formation of the complete, native [(EutB–EutC)₂]₃ oligomer, is verified by biochemical methods. Prediction that cysteine residues on the exposed subunit–subunit contact surfaces of isolated EutB and EutC will interfere with assembly by cysteine formation is verified by activating effects of disulfide reducing agents. Angstrom-scale congruence of the reconstituted and native EAL in the active site region is shown by electron paramagnetic resonance spectroscopy. Overall, the hierarchy of subunit interactions and microscopic features of the contact surfaces, which are revealed by the homology model, guide and provide a rationale for a refined genetic and biochemical approach to reconstitution of the functional [(EutB–EutC)₂]₃ EAL oligomer. The results establish a platform for further advances in understanding the molecular mechanism of EAL catalysis and for insights into therapy-targeted manipulation of the bacterial *eut* metabolosome.



Ethanolamine ammonia-lyase (EAL)^{1,2} is a 5'-deoxyadenosylcobalamin [AdoCbl, coenzyme B₁₂ (Scheme 1)]-dependent bacterial enzyme [EC 4.3.1.7; cobalamin (vitamin B₁₂)-dependent enzyme superfamily]^{3,4–6} that catalyzes the deamination of the short-chain vicinal amino alcohols, aminoethanol and (S)- and (R)-2-aminopropanol.⁷ The coding sequence for EAL is located on the 17-gene *eut* operon,⁸ which contains coding regions for the proteins required for bacterial utilization of ethanolamine as a carbon and nitrogen source, in the presence of exogenous AdoCbl.^{9–11} Studies of the *eut* metabolosome suborganellar structure have focused on our molecular understanding of the hierarchical assembly,¹² and the complex chemistry that is involved in ethanolamine processing.¹³ Ethanolamine utilization has been identified as a key competitive advantage for colonization of the mammalian gut by bacterial pathogens, and ethanolamine can serve as a trigger for virulence.^{14,15} A strong correlation has been found between food-poisoning bacteria and the presence of *eut* genes.¹⁶

EAL, the central enzyme in ethanolamine utilization, is a “radical enzyme” that harnesses the high reactivity of electron-

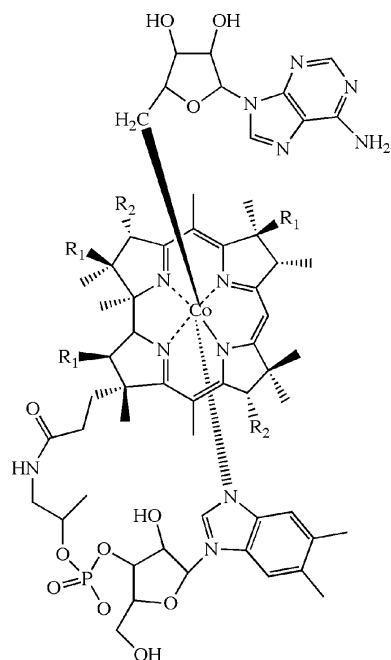
deficient species to perform catalysis.¹⁷ Previous detailed studies of EAL structure and mechanism have elucidated the geometry of the reactant centers in the active site,^{18–21} and determined the kinetics of culled steps in the reaction cycle of EAL,^{22–26} in the native enzyme from *Salmonella typhimurium*. X-ray crystallographic structures of the complete EAL from a different bacterium, *Escherichia coli*, including the bound cofactor and substrate have been obtained,²⁷ supplanting an earlier incomplete structure of EutB from *Listeria monocytogenes* (PDB entry 2QEZ) and a model for *S. typhimurium* EutB based on secondary structure prediction.²⁸ Further advances in elucidating the molecular mechanism of EAL require a high-resolution model for the *S. typhimurium* protein, for structure–function correlations, and the development of methods for efficient site-directed mutagenesis of EAL. Progress has been thwarted by the large mass (12 subunits, ~500 000 g/mol) and

Received: December 11, 2012

Revised: January 30, 2013

Published: February 1, 2013

Scheme 1. Structure of 5'-Deoxyadenosylcobalamin



low solubility (<2 mg/mL) of the EAL oligomer,²⁹ and individual subunit instability. Here, we present a robust, homology-modeled structure of EAL from *S. typhimurium*, which is based on the *E. coli* structure, and use it to develop and rationalize approaches for expression and reconstitution of functional EAL from purified, individual protein subunits.

EAL is composed of two subunits, encoded by the *eutB* and *eutC* genes. The larger subunit, EutB (453 residues, 49.4 kDa, *S. typhimurium*), is a ($\beta\alpha$)₈ TIM-barrel protein³⁰ that contains the substrate binding cavity at the C-terminal end of the β -barrel.²⁸ The substrate binding cavity is capped by the AdoCbl binding site. The cofactor site and C-terminal end of the β -barrel are covered by the smaller subunit, EutC (298 residues, 32.0 kDa, *S. typhimurium*).²⁷ The EutB–EutC heterodimer is the fundamental structural unit of EAL. In the EutB₆EutC₆ EAL oligomer (488 kDa, *S. typhimurium*), two EutB–EutC heterodimers are organized as a (EutB–EutC)₂ homodimer, and three homodimers are organized into a [(EutB–EutC)₂]₃ trimer. The native *eutB* and *eutC* genes are under control of the same promoter and are co-expressed in vivo.

The high molecular mass of the EAL oligomer and the associated low solubility²⁹ are linked to the quasi-solid state of the enzyme in the *eut* metabolosome suborganelle, in vivo. In vitro, the strong aggregation propensity interferes with expression and purification yields of correctly folded, functional EAL (because of inclusion body formation) during expression of plasmids that include the cloned *eutB-eutC* coding region, in *E. coli* vectors. Techniques employed for growth of bacteria in which EAL is overexpressed have used systems designed to lower the rate of gene expression, as follows: (a) growth and expression under noninducing conditions,³¹ (b) use of the BD21 pLysS system,³² and (c) low-temperature growth.²⁷ Previous attempts to use the alternative approach of reassembling functional EAL from its separately expressed and purified subunits, EutB and EutC, have been unable to recover activity.²⁹ This may, in part, be related to the observed instability of isolated EutC. The Instability Index³³ of EutC is 51.4 (a score of >40 connotes high instability), compared with

a value of 27.0 for EutB. Consistent with these predictions, EutC is subject to rapid trypsinization.³⁴ In addition, the N-terminal region of EutC may drive the formation of inactive EAL aggregates.³⁴

We generate a multiscale homology model structure of *S. typhimurium* EAL, which describes the molecular structure of the subunits, including the active site interactions of side chains, the substrate, and the cofactor, and the global subunit interactions in the oligomer, that is based on the X-ray crystallographic structure of *E. coli* EAL.²⁷ The homology model predicts that a EutB hexamer can form in the absence of EutC, and that cystine disulfide bond formation from cysteine sulfhydryl groups exposed in the isolated subunits will promote protein aggregation. Model predictions in concert with biochemical manipulations lead to the development of a low-temperature growth and expression system in NEB T7 Express I⁹ *E. coli*, which allows the individual expression of soluble EutB and EutC. Using tandem affinity and size exclusion purification, the EutB subunit can be purified to homogeneity, and EutC can be enriched to >95%. Judicious application of sulfhydryl reagents allows the reconstitution of active EAL.

Overall, the EAL structural model predictions inform the development of a novel sequential assembly protocol for reconstitution of the functional *S. typhimurium* EAL oligomer from individual subunits, which is verified by biochemical, spectroscopic, and functional studies.

MATERIALS AND METHODS

Homology Modeling of *S. typhimurium* EAL. EAL sequence alignment was performed with ClustalW2^{35,36} to map the *S. typhimurium* EAL sequence onto the *E. coli* EAL sequence. The complete alignment is given in Figure S1 of the Supporting Information. Residues 4–43 of the N-terminal end of the EutC subunit of the *E. coli* EAL were truncated for crystallization.²⁷ Therefore, these residues are not present in the X-ray crystallographic structure.²⁷ As a result, the modeled structure also does not contain the first 46 residues of EutC. Five homology models of the [(EutB–EutC)₂]₃ functional oligomeric structure were generated by using this alignment and the *E. coli* crystal structure of PDB entry 3ABO with MODELER.^{27,37} MODELER generates homology-modeled protein structures by using comparative modeling based on spatial restraints. MODELER minimizes an effective energy function, which is based not only on alignment with the template structure but also on statistical experimental restraints from similar structural features, such as main-chain bond lengths and dihedral angles from equivalent groups. In addition to these homology-derived restraints, MODELER also takes into account stereochemical restraints using the CHARMM22 force field.

The models generated were first assessed by using the Discrete Optimized Protein Energy (DOPE) profile.³⁸ The DOPE profile allows quick evaluation of a putative structure by distinguishing high-energy regions within a structure, which may indicate possible errors. However, it is not an absolute measure and is only useful for ranking models generated from the same alignment. The structure with the lowest DOPE score was selected for further analysis. The modeled structure was validated by using the PROCHECK test suite.³⁹ PROCHECK tests the stereochemical quality of protein structures, by assessing main-chain parameters such as bond lengths, bond planarity, and ω , ϕ , and ψ bond angles, in addition to side-chain parameters such as χ_1 – χ_4 bond angles. The structure was

visually examined by using SYBYL version 8.0 (Tripos Inc., St. Louis, MO) and PyMOL (DeLano Scientific LLC, San Francisco, CA).

The solvent accessibility of residues in the modeled structure was judged using the ASAView web server.⁴⁰ Briefly, ASAView uses the DSSP program⁴¹ to calculate the surface area of each amino acid residue that is accessible to solvent and scales this solvent accessible surface area by the total area of that residue. Relative solvent accessibilities for EutB and EutC monomers, the EutB–EutC heterodimer, the (EutB–EutC)₂ homodimer, and the [(EutB–EutC)₂]₃ oligomer were computed and compared to assess the changes upon hexamer formation.

The electrostatics of the modeled protein surface were calculated by using the Adaptive Poisson–Boltzmann Solver, APBS.⁴² The PDB2PQR web server,^{43,44} a tool that automates the conversion of protein structural coordinates (PDB entry) into files containing the calculated charge information (PQR), was used. PDB2PQR builds and assesses the hydrogen bonding network of the protein structure and estimates protonation states, charges, and atomic radii for the biomolecule. APBS uses this input information to solve the Poisson–Boltzmann equation in a parallel manner, allowing the evaluation of the electrostatic potentials for the modeled structure. The results were visualized in PyMOL.

Protein Library. A search of the Pfam database⁴⁵ yielded 693 sequences for EutB and 689 for EutC. A multiple-sequence alignment of these sequences was downloaded, and examined with the MATLAB Bioinformatics Toolbox (Mathworks Inc., Natick, MA). In this alignment, EutB residues 11–452 and EutC residues 63–297 of *S. typhimurium* were aligned. At each position in the alignment, the most frequent residue was used to build a consensus sequence, which was used to assess the degree of conservation.

Construction of the EALH6 Plasmid. The 8.5 kb recombinant wild-type DNA encoding the EutB and EutC subunits of EAL from *S. typhimurium* in the pBR322 vector was extracted from the *E. coli* overexpression strain.³¹ Primers encoding an N-terminal His₆ tag complementary to the *eutB* starting sequence were obtained from Invitrogen (Life Technologies Corp., Carlsbad, CA), as follows: EutBH6 FP, GGG GAA CGA CTT ATG CAT CAC CAT CAC CAT CAC AAA CTA AAG ACC ACA TTG; EutBH6 BP, CAA TGT GGT CTT TAG TTT GTG ATG GTG ATG GTG ATG CAT AAG TCG TTC CCC.

Mutagenesis was conducted using the QuikChange II XL site-directed mutagenesis kit from Stratagene (La Jolla, CA) according to the manufacturer's instructions. Briefly, the wild-type EAL plasmid was used as a template for polymerase chain reaction (PCR) amplification of the His₆ primers with high-fidelity *PfuUltra* DNA polymerase. The endonuclease, *DpnI*, was used to digest the methylated template DNA, leaving the new EALH6 DNA intact. The new plasmid was then transformed into XL10 Gold ultracompetent cells (Stratagene), and the sequence was verified by DNA sequencing (performed by Beckman Coulter Genomics, Danvers, MA).

Construction of EutBH6 and EutCH6 Plasmids.
Preparation of the pET28a Plasmid. An ampicillin resistant pET28a plasmid with a His₆ tag was used. The plasmid was digested in the multiple-cloning site polylinker with the restriction enzymes *XhoI* and *NdeI* from New England Biolabs (Ipswich, MA), and the 5 kb cleaved sticky-ended DNA fragment was purified from a 1% agarose Tris-acetate-EDTA

(TAE) gel by using the QIAquick gel extraction kit (QIAGEN, Venlo, The Netherlands).

Preparation of the EAL Gene. The wild-type EAL pBR322 sequences did not contain any *NdeI* (CA'T ATG) or *XhoI* (C'TC GAG) restriction sites. Primers were purchased from Integrated DNA Technologies (Coralville, IA) to insert these restriction sites at the start and end of the EutB and EutC genes separately, as follows: EutBNdeI FP, AAA AAA AAA A CAT ATG AAA CTA AAG ACC ACA TTG TTC GGC AAT G; EutBXhoI BP, AAA AAA AAA A CTC GAG TCA GAA GAA CAA TGA CGG ATC GCC CGC CCG TTT GG; EutCNdeI FP, AAA AAA AAA A CAT ATG GAT CAA AAA CAG ATT GAA GAA ATT GTA CG; EutCXhoI BP, AAA AAA AAA A CTC GAG TTA ACG GGT CAT GTT GAT GCC GGA CGC TTT. PCR was performed with *Pfu* DNA polymerase, and the *eutB* (1.5 kbp) and *eutC* (0.9 kbp) genes were purified from the resulting sample by electrophoresis on a 1% agarose TAE gel. DNA was extracted from the gel and then digested with the restriction enzymes *XhoI* and *NdeI*, yielding sticky ends complementary to the pET28a plasmid. These sticky-ended genes were purified again using agarose gel electrophoresis.

Construction of the Final Plasmids. Sticky-ended pET28a and *eutB* or *eutC* were ligated by using T4 DNA ligase. These constructs were transformed into XL10 Gold ultracompetent cells. The plasmid was then transferred into T7 Express I^q competent cells (New England Biolabs), because these cells were found to be optimal for gene expression. The sequences of the EutB and EutC transformants were verified by Beckman Coulter Genomics and were found to be 100% identical between the restriction sites.

Bacterial Growth and Protein Purification. *Growth.* Cells were grown from glycerol stocks in 5 mL of 2× YT starter cultures with 100 µg/mL ampicillin at 37 °C (30 °C for EutB and EutC) overnight while being shaken. This was used to inoculate 500 mL of medium that was allowed to grow until the OD₆₀₀ reached ~0.8. Following induction by 0.4 mM IPTG, cells were grown for 5 h and then harvested by centrifugation at 3000g and 4 °C. Cells were washed with 10 mM potassium phosphate (KP_i) buffer (pH 7.5), and the pellet was frozen in liquid nitrogen and stored at –80 °C.

Purification. Cells were thawed and resuspended in 5 mL of lysis buffer [50 mM NaH₂PO₄, 300 mM NaCl, and 10 mM imidazole (pH 8.0)], which was supplemented with 1 mg/mL lysozyme, 50 µL 100× HALT protease inhibitor cocktail (ThermoFisher Scientific, Waltham, MA), and 2 µL of Benzonase Nuclease (Sigma-Aldrich, St. Louis, MO) per gram of wet weight. The cells were then lysed on ice for 1 h and sonicated at 10 W (10 s on–off cycle) for 3 min. The cell debris was separated by centrifugation at 10000g and 4 °C for 20 min. The supernatant was filtered with a 0.45 µm filter and then loaded onto a 5 mL HisTrap Fast Flow crude column that was mounted in an AKTA Purifier fast protein liquid chromatography (FPLC) system (GE Healthcare Life Sciences, Schenectady, NY). The column was washed with lysis buffer, followed by elution buffer (lysis buffer with 250 mM imidazole added) in a 1:9 ratio with lysis buffer (34 mM imidazole). EAL was then eluted using 40% elution buffer (106 mM imidazole). At higher elution buffer percentages, the EAL in the eluate was deficient in EutC. For EutBH6 and EutCH6, elution was performed at 100% elution buffer.

The protein-containing fraction (10 mL) was dialyzed in 10 K membrane dialysis tubing against 1 L of 10 mM KP_i buffer

(pH 7.5) for 6 h, followed by buffer replacement, and dialysis overnight at 4 °C. The protein was then concentrated using aquacide for 1 h and spin-concentrated using a Pierce 7 mL concentrator with a 10 K membrane (ThermoFisher Scientific). The resulting protein concentration was checked using the Bradford protein assay,⁴⁶ by using a 2 mg/mL bovine serum albumin standard (ThermoFisher Scientific). For purified EutB and EutC, the concentration was directly assessed using ultraviolet absorption spectroscopy, by using the sequence-predicted extinction coefficients⁴⁷ at 280 nm of 33350 and 18450 M⁻¹ cm⁻¹, respectively. Absorption spectroscopy measurements were performed on a Shimadzu UV-1601 absorption spectrophotometer.

Gel Filtration. Gel filtration was performed on an analytical Superdex 200 10/300 GL column from GE Healthcare Life Sciences by using the AKTA Purifier FPLC system. A 100 μ L volume of protein was injected into the column, which was pre-equilibrated with 10 mM KP_i buffer (pH 7.5), and run at a flow rate of 0.5 mL/min. Absorbance was monitored at 280 nm.

SDS and Native PAGE. Polyacrylamide gel electrophoresis (PAGE) was performed by using 10% Mini Protean TGX gels from Bio-Rad Laboratories (Hercules, CA) according to known protocols.^{48,49} Coomassie Brilliant Blue dye was used for staining.

Enzyme Activity Assay. Enzyme activity was determined by using the coupled alcohol dehydrogenase assay, essentially as described previously.⁵⁰ The assay mixture contained 10 mM KP_i buffer (pH 7.5), a 2:1 5'-deoxyadenosylcobalamin/active sites mixture, substrate, 120 μ M NADH, and 4.3 units of alcohol dehydrogenase at 25 °C. For EAL reconstitution, a 2:1 5'-deoxyadenosylcobalamin/active sites mixture was added to EutB in a 10 mM KP_i buffer (pH 7.5) with 5 mM reductant, while EutC was incubated separately. Incubations prior to the assay were performed for 1 h on ice. The two subunits were mixed, and enzyme activity was assayed, as described above.

Electron Paramagnetic Resonance Spectroscopy. Samples for EPR spectroscopy were prepared under dim red-safe lighting and protected from light during the experiment to prevent photodegradation of 5'-deoxyadenosylcobalamin. All preparations were performed on ice. A 2:1 5'-deoxyadenosylcobalamin/active sites mixture was added to 60 μ M enzyme (active site concentration; corresponding EAL concentration of 10 μ M) in 10 mM KP_i buffer (pH 7.5) with 5 mM TCEP. (S)-2-Aminopropanol (30 mM) was added to this mixture to start the reaction. The sample was mixed, transferred to a 2 mm outside diameter quartz EPR tube, and flash-frozen in liquid nitrogen. Less than 15 s elapsed between mixing and freezing. EPR spectra were obtained by using a Bruker E500 ElexSys EPR spectrometer that was equipped with a Bruker ER4123 SHQE cavity. The temperature was controlled by using a Bruker ER4131VT liquid nitrogen/gas flow cryostat system. All spectra were collected at a microwave power of 2 mW and microwave frequency of 9.45 GHz, at 120 K.

RESULTS AND DISCUSSION

Homology Modeling of *S. typhimurium* EAL. Forms of EAL from *E. coli* and *S. typhimurium* are 98% identical in sequence in the EutB subunit and 91% in the EutC subunit (Figure S1 of the Supporting Information).^{27,31} In EutB, there are nine nonconserved positions (Figure S2 of the Supporting Information). Two of these positions, EutB322 [I (*E. coli*) \rightarrow L (*S. typhimurium*)] and EutB159 [A (*E. coli*) \rightarrow C (*S. typhimurium*)], are on the interior β -strands of the ($\beta\alpha$)₈ barrel

structure. These residues lie near the N-terminal end of the barrel and are not predicted to make substrate contacts. Two other nonconserved positions, EutB203 [S (*E. coli*) \rightarrow T (*S. typhimurium*)] and EutB209 [I (*E. coli*) \rightarrow V (*S. typhimurium*)], are on the α -helices on the exterior of the barrel, and another position, EutB185 [V (*E. coli*) \rightarrow A (*S. typhimurium*)], is in the loop region that connects barrel α -helix 1 to β -strand 2 at the N-terminal end of the barrel. EutB87 [Q (*E. coli*) \rightarrow R (*S. typhimurium*)] is involved in the binding interface between EutB subunits. There are a larger number, 26, of nonconserved residues in EutC. Of these, only EutC78 [V (*E. coli*) \rightarrow A (*S. typhimurium*)] is involved in the EutC–EutB binding interface. The high degree of sequence identity allowed *S. typhimurium* EAL to be modeled directly from the template structure of the *E. coli* enzyme,²⁷ following sequence alignment.^{35,36} The selected template was the X-ray crystallographic structure of EAL (PDB entry 3ABO). The 3ABO structure includes the complete [(EutB–EutC)₂]₃ oligomer structure, and the EutB–EutC dimer units contain the bound substrate, ethanolamine, and the cobalamin cofactor analogue, cyanocobalamin. Five homology models were generated using MODELER.⁵¹ The models were analyzed by using the Discrete Optimized Protein Energy (DOPE), which is an effective energy function that allows comparison between modeled structures.³⁸ The model with the lowest DOPE score was selected. The selected model was also validated by using the PROCHECK model stereochemistry test suite (Figure S3 of the Supporting Information).³⁹ The model showed 99.9% of residues in allowed regions of the ϕ , ψ , or Ramachandran, plot, of which 93.5% were in favored regions. The two residues that are in disallowed regions [EutC(L197) and EutC(S156)] are present in loop segments outside of the α - and β -secondary structure and therefore do not have a significant impact on the structure of the model. The model structure was also assessed by using the G factor, a log-odds score based on observed distributions³² of stereochemical parameters. Values of the G factor were all significantly larger than the limits of -0.5 (probability of being incorrect) (Table S1 of the Supporting Information).

The homology model of the *S. typhimurium* EAL oligomer structure is presented in Figure 1. An overlay of the homology model and *E. coli* secondary structures is shown in Figure S4 of the Supporting Information. The positions of nonhomologous residues are highlighted in Figure S5 of the Supporting Information, and the root-mean-square deviation of backbone C α atoms is presented in Figure S6 of the Supporting Information.

Subunit Interaction Hierarchy from the Homology Model of *S. typhimurium* EAL. Figure 1 illustrates the different hierarchies of organization of the EutB and EutC subunits in the oligomer. (a) EutB and EutC monomers form the EutB–EutC heterodimer. (b) Two heterodimers form a (EutB–EutC)₂ homodimer. (c) Three (EutB–EutC)₂ homodimers form the biologically active oligomer, which is described as a trimer of dimers, [(EutB–EutC)₂]₃. Thus, there are two different types of contact faces between EutB–EutC units. The (EutB–EutC)₂ intrahomodimer contact face has an area of 1293 Å² and is shown in Figure 1B. The edge-to-edge contact from interhomodimer formation of the trimer, shown in Figure 1C, has a smaller surface area of 804 Å². The primary contacts for both the intra- and interhomodimer interfaces are on the EutB subunits. The model shows that the EutC subunit is not involved in the edge-to-edge, intrahomodimer interaction.

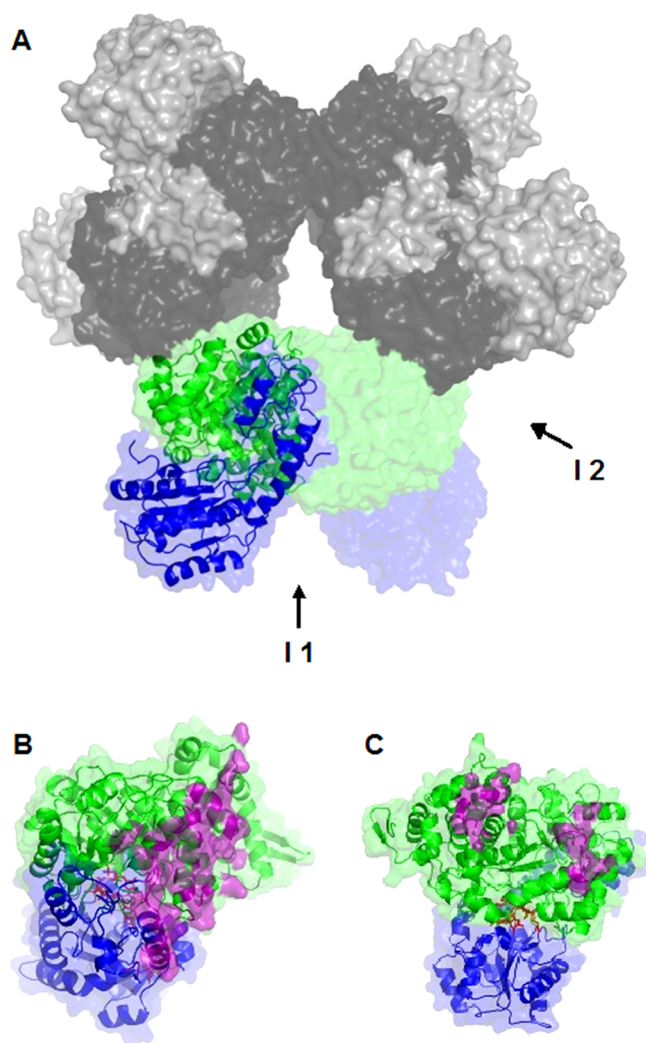


Figure 1. Surface and secondary structure representations of the homology-modeled structure of *S. typhimurium* EAL. (A) [(EutB–EutC)₂]₃ EAL oligomer. One (EutB–EutC)₂ heterodimer is colored green (EutB subunits) and blue (EutC subunits), and two other heterodimers are colored in shades of gray (EutB, dark gray; EutC, light gray). (B and C) Two EutB–EutC heterodimers that compose one (EutB–EutC)₂ homodimer. EutB is colored green and EutC blue. (B) Intraheterodimer interface (denoted I1, in panel A), with residues involved in the contact surface colored purple. (C) Interheterodimer interface (denoted I2, in panel A), with residues involved in the contact surface colored purple.

Therefore, the model predicts that the hexameric (EutB₂)₃ trimer will form in solution, in the absence of EutC.

Figure 2 shows the common contact face between EutB and EutC, which has a surface area of 2080 Å². The EutC contact face has a higher incidence of positively charged residues (23% positive, 10% negative), compared with the corresponding face on EutB, which carries the opposite charge (6% positive, 20% negative), where the percentages are relative to all contact residues. Figures S7 and S8 (Supporting Information) present the classes of contact surface residues and the location in EutB and EutC, respectively. In addition, two salt bridges are identified in this binding interface: EutB(D398)–EutC(R80) and EutB(D411)–EutC(R89). These results suggest that static charge–charge interactions contribute significantly to EutB–EutC heterodimer assembly and, therefore, that low-ionic strength conditions will favor intersubunit interactions.

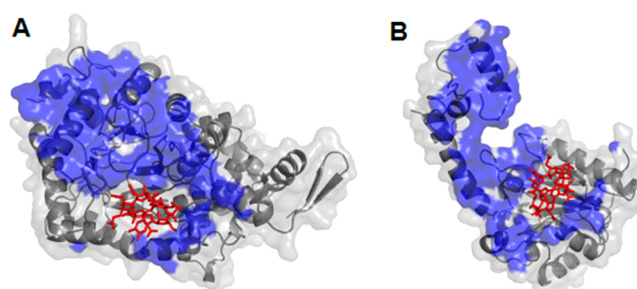


Figure 2. Contact surfaces of the EutB and EutC subunits in the EutB–EutC heterodimer, in the modeled structure of *S. typhimurium* EAL: (A) EutB subunit and (B) EutC subunit. The contact face is shown as a blue surface. The 5'-deoxyadenosylcobalamin cofactor is represented as red sticks.

The EAL from *S. typhimurium* contains 16 cysteine residues. Eight of these cysteine residues (C223, C361, C363, C364, and C388 in EutB and C158, C243, and C261 in EutC) are highly conserved, based on a survey of 693 EutB sequences and 689 EutC sequences from different bacterial species and strains.⁴⁵ Figure 3 illustrates the positions of the cysteine residues in the

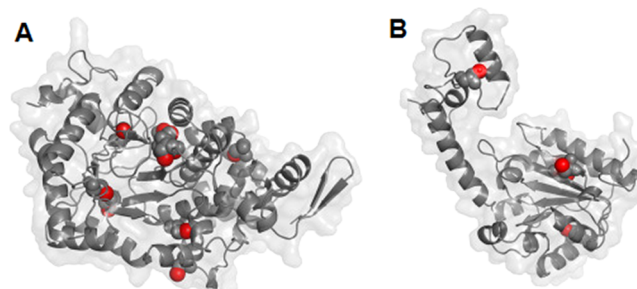


Figure 3. Positions of cysteine side-chain sulfhydryl groups (red spheres) in the EutB and EutC subunits in the modeled structure of *S. typhimurium* EAL: (A) EutB subunit and (B) EutC subunit. The subunit orientations are the same as those shown in Figure 2.

EutB and EutC structures. The conserved residues include an unusual CDC motif, in the active site region in EutB (residues C361, C363, and C364), and C261 of EutC, which is in the cofactor binding site. Both of these active site regions become solvent-exposed in the isolated subunits and, therefore, during the separate expression of EutB and EutC. Native EAL appears to be devoid of cystine disulfide bonds, with the possible exception of cysteines in the active site, and C37 and C46 of EutC, which were not modeled, because they are present on a cleaved leader sequence in *E. coli*. However, the EAL homology model indicates that, when the subunits are grown separately, opportunistic disulfide linkages may form, which indicates a requirement for a disulfide-reducing environment, during the reconstitution of isolated EutB and EutC subunits.

Expression and Purification of Wild-Type EAL. The His₆-tagged, wild-type EAL construct, EALH6, allows expression of EAL at 37 °C without significant inclusion body formation. An SDS–PAGE gel of overexpressed EAL purified from this construct is shown in Figure 4. The EutB and EutC protein subunits have calculated masses of 48 and 30 kDa, respectively, which are comparable to the expected masses of 50 and 32 kDa, respectively. The differences in masses between EutB and EutC from the expressed EAL oligomer and the individually purified subunits are explained by the presence of

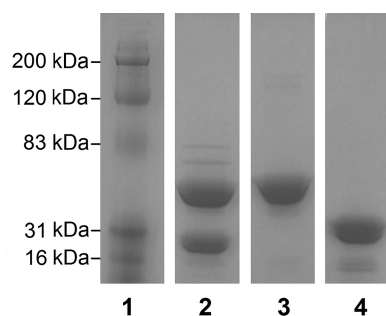


Figure 4. Sodium dodecyl sulfate–polyacrylamide gel electrophoresis of purified proteins. Lanes contained (1) the protein molecular mass ladder, (2) EAL purified from the EALH6 construct, (3) EutB, and (4) EutC. Lane 4 was adjusted to correct for dye front curvature.

the pET28 His₆ tag sequence in the plasmids used for the purification of the individual protein subunits.

EAL purified from the EALH6 construct had a higher proportion of the EutB protein, relative to EutC. This phenomenon has been previously noted in other EAL overexpression systems and attributed to more efficient expression of genes that are closer to the promoter region, relative to genes further downstream.³⁴ In the EALH6 pBR322-based construct, *eutB* precedes *eutC* and is therefore closer to the *lac* promoter. Alternative explanations are that incompletely constructed EAL has a higher affinity for the nickel column than the complete oligomer, or that EutC is released from column-bound EutB during the washing procedures. To accomplish the purification of the homogeneous [(EutB–EutC)₂]₃ complex, we performed elution from the nickel affinity column at a relatively low imidazole concentration (100 mM). At higher imidazole concentrations (250 mM), the eluted EAL was deficient in EutC (Figure S9 of the Supporting Information). Densitometric analysis⁵³ of SDS–PAGE gels showed that the subunit molar ratio of the purified EAL was 1.00:1.01, following elution with 100 mM imidazole, compared to a molar ratio of 1.00:0.32 for elution with 250 mM imidazole (Table S2 of the Supporting Information). In addition, native PAGE and gel filtration experiments show no evidence of excess EutB or EutC in purified EALH6 (Figure 5 and Figure S10 of the Supporting Information). In summary, the results indicate the following. (a) EutC-deficient EutB–EutC oligomer-like complexes bind with higher affinity to the nickel affinity column. (b) The purified [(EutB–EutC)₂]₃ complex can be obtained by elution with 100 mM imidazole.

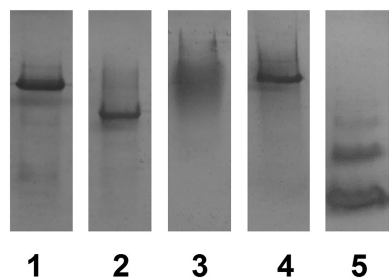


Figure 5. Native polyacrylamide gel electrophoresis of purified proteins. Lanes contained (1) EAL purified from the EALH6 construct, (2) EutB, (3) EutC, (4) EAL reconstituted from the purified EutB and EutC subunits, and (5) the bovine serum albumin standard. Lane 5 was adjusted to correct for dye front curvature.

Expression and Purification of Individual EutB and EutC Proteins.

The formation of insoluble aggregates during the separate, individual expression of EutB and EutC proteins from the respective constructs was prevented by expression at a low temperature (30 °C) in NEB T7 Express I^q *E. coli* cells. The tight control of EAL expression by the upregulated *lac* repressor in this strain⁵⁴ prevents the formation of inclusion bodies and leads to the formation of soluble EutB and EutC. Using a single step of nickel affinity column chromatography, >90% pure protein could be obtained as judged by densitometric analysis of SDS–PAGE gels. The EutB and EutC proteins were purified further, by using size exclusion chromatography. EutB was purified to homogeneity, and EutC was purified to >95%. An overloaded SDS–PAGE gel of the purified protein preparations is shown in Figure 4. The mass of the EutB protein was calculated to be 50 kDa, which agrees well with the sequence-predicted mass of 52 kDa. The EutC protein had a calculated mass of 33 kDa, which is comparable to the sequence-predicted value of 34 kDa. Following purification, the EutC protein showed visible aggregation over the course of 2 h, when at high concentrations (>5 mg/mL). Therefore, care was taken to ensure that EutC was not kept at high concentrations for periods of >15 min. The EutB protein was not observed to form higher-molecular mass aggregates, up to a protein concentration of 40 mg/mL, over the course of 2 h. The results show that the individual EutB and EutC proteins can be expressed and purified, separately, by the developed protocol.

Characterization of the Oligomeric State of Individually Expressed and Purified EutB and EutC.

Figure 5 shows nondenaturing PAGE results for the protein purified from expression of the EALH6, EutB, and EutC constructs. The EALH6 protein oligomer (494 kDa, lane 1) migrates as a single dominant band, with a relative mobility (*R_f*) of 0.187. The purified EutB protein (Figure 5, lane 2) displays a single dominant band with a larger *R_f* value (0.247). The presence of a single band for purified EutB is consistent with the presence of EutB as the 310 kDa (EutB₂)₃, or EutB₆ hexameric, oligomer. This interpretation is supported by gel filtration experiments (Figure S10 of the Supporting Information), which show that purified EutB elutes closer to the authentic EALH6 oligomer (+1.0 mL) than to purified EutC (+5.5 mL). Although differences in protein size, structure, and charge influence *R_f* values in nondenaturing PAGE, so that comparisons of different proteins are approximate, at best, we note that the bovine serum albumin control in Figure 5 (lane 5) shows characteristic monomer (66 kDa), dimer (132 kDa), and trimer (198 kDa; *R_f* = 0.265) bands,⁵⁵ which display relatively large *R_f* values.

Figure 5 also shows that the purified EutC subunit migrates as a diffuse band in the nondenaturing gel. Under denaturing and reducing conditions, EutC migrates as a single band via SDS–PAGE, as shown in Figure 4, and under reducing conditions, it elutes in a single fraction via gel filtration (Figure S10 of the Supporting Information). Therefore, the diffuse band on the nondenaturing gel is caused by nonspecific aggregation of isolated EutC. The first 27 residues of EutC have been associated with the low solubility and aggregation of wild-type EAL in *E. coli*.³⁴ It was found that these residues are not essential for activity and can be removed.³⁴ The first 19 residues of *E. coli* EutC are homologous to those in *S. typhimurium*, and have been identified as a signaling sequence that targets EAL to the eut metabolosome microcompartment, where EAL is found in vivo.⁵⁶ The residues in the N-terminal region of EutC may

induce the formation of aggregates of varying sizes and conformations.

Assembly of EAL from Individually Expressed and Purified EutB and EutC. Figure 5 shows the nondenaturing PAGE results for reconstitution of 1:1 (molar ratio) purified EutB and EutC proteins. The reconstituted EAL (lane 4) has an R_f of 0.178, compared with an R_f of 0.187 for wild-type EAL (lane 1). The closely similar R_f values indicate that the [(EutB–EutC)₂]₃ EAL oligomer is reconstituted from its component subunits. The reconstituted EAL shows some evidence of diffuse staining at R_f values less than the principal band. This may represent improperly constructed EAL, because of aggregation, because EutB is present in stoichiometric amounts with EutC. In summary, the results demonstrate that the developed purification and reconstitution protocols lead to the native [(EutB–EutC)₂]₃ oligomeric structure of EAL, from isolated EutB and EutC subunits.

Steady-State Enzyme Activity of EAL Reconstituted from Isolated EutB and EutC. The homology model structures in Figure 3 show that cysteine residues are exposed on the contact surfaces of both the EutB and EutC subunits. Thus, EutB and EutC subunits, which are expressed and purified individually, will have solvent-exposed cysteine residues in the active site region, and in the absence of the reducing intracellular environment,⁵⁷ these cysteines may undergo oxidative reactions to form intra- and/or intersubunit disulfide bonds. Consistent with this prediction, cystine disulfide bond reducing agents were found to be crucial for the functional reconstitution of EAL from isolated EutB and EutC subunits, as shown by the k_{cat} values for different reconstitution conditions in Figure 6. A k_{cat} value of $0.51 \pm 0.09 \text{ s}^{-1}$ is obtained, when no cystine disulfide reducing agent is added to the postaffinity

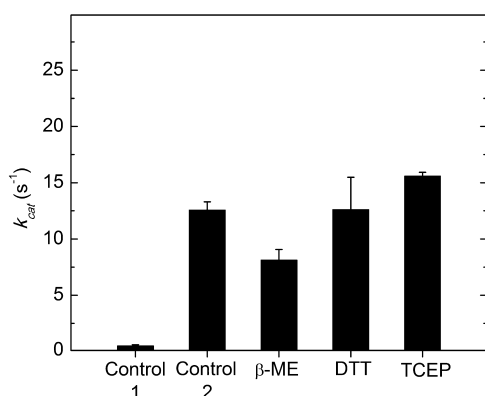


Figure 6. Dependence of the activity of reconstituted EAL on the presence of different disulfide reducing agents. Activity represents the maximal turnover number (k_{cat}) attained with excess ethanolamine substrate (concentration $\gg K_M$), under the assay conditions described in Materials and Methods. The EutB and EutC proteins were present at stoichiometric concentrations. The bars correspond to the following conditions. In Control 1, the reductant was omitted during the purification procedure and during activity measurement. All other samples were purified in the presence of 5 mM TCEP reductant, which was removed as a consequence of the gel filtration step of the purification procedure. In Control 2, no reductant was added to the kinetic assay mixture. In β -ME, β -mercaptoethanol (5 mM) was added to the kinetic assay mixture. In DTT, dithiothreitol (5 mM) was added to the assay mixture. In TCEP, Tris-2-carboxyethyl phosphine (5 mM) was added to the assay mixture. Values represent the averages and standard deviation (error bars) of three measurements.

column eluate, prior to the sample being loaded onto the gel filtration column, and when no cystine disulfide reducing agent is present in the kinetic assay buffer. Figure 6 shows that addition of the disulfide reducing agent, tris(2-carboxyethyl)-phosphine (TCEP), to the sample, prior to loading on the gel filtration column, leads to a 25-fold increase in the k_{cat} value to $12.6 \pm 0.7 \text{ s}^{-1}$. An additional increment to $15.6 \pm 0.3 \text{ s}^{-1}$ is achieved when TCEP is added to the separate, isolated EutB and EutC and the mixture is incubated for 2–5 min, prior to the mixing of the subunits for the steady-state kinetic assay. Figure 6 shows that preincubation with TCEP prior to the steady-state kinetic assay mixture is more effective in increasing k_{cat} than the disulfide reducing agents, β -mercaptoethanol (β -ME) and D,L-dithiothreitol (DTT) at equivalent concentrations. β -ME is a monothiol reductant and is therefore less effective than the dithiol, DTT.^{58,59} From the selectivity of TCEP for reducing cysteine residues in the protein,⁶⁰ we can conclude that the primary mechanism of inactivation of EAL is oxidation of cysteines, rather than other pathways, such as proteolytic degradation.

Although the enhancement of k_{cat} by DTT is comparable to that caused by TCEP, within the standard deviation of the activity measurements, the stability of DTT is reduced by nickel contamination from the affinity purification step.⁵⁹ In addition, TCEP is compatible with sulfhydryl-specific spin labeling reagents at the TCEP concentration used,⁶¹ which is a consideration for downstream EPR spectroscopy-associated experiments. Therefore, TCEP was selected as the cystine disulfide reductant for use in the purification and manipulation of EutB and EutC.

Table 1 shows the steady-state enzyme kinetic parameters for the EutB- and EutC-reconstituted EAL and the His₆-tagged

Table 1. Steady-State Enzyme Kinetic Parameters for the Purified EAL Oligomer and EAL Oligomer Reconstituted from EutB and EutC Subunits^a

	k_{cat} (s^{-1})	K_M (μM)	k_{cat}/K_M ($\text{M}^{-1} \text{s}^{-1}$)
EAL	22.7 ± 0.3	31.6 ± 4.5	$(7.2 \pm 1.1) \times 10^5$
reconstituted EAL	16.0 ± 2.4	22.0 ± 4.2	$(7.3 \pm 2.5) \times 10^5$

^aResults represent the average and standard deviation for at least three measurements.

EALH6 oligomers. The k_{cat} value of reconstituted EAL is $70 \pm 16\%$ of the k_{cat} for EALH6, and the K_M value was reduced by 1.4-fold. The difference in K_M between EALH6 and reconstituted EAL may be caused by a subtle effect on the protein structure that arises from the different His₆ tags. In EALH6, there is one His₆ tag, which is located on the EutB subunit. The individual EutB and EutC subunits each have a His₆ tag, which is adjacent to a linker region. Thus, the reconstituted EAL includes two His₆ tags. The reduction in k_{cat} is consistent with the level of functional reconstitution from isolated subunits, as described in more detail below. The catalytic efficiency, or “specificity constant”,⁶² which is given by k_{cat}/K_M , does not differ significantly for the two EAL oligomers. Isolated, purified EutB and EutC showed no detectable activity ($<0.03 \text{ s}^{-1}$). In summary, the steady-state enzyme kinetic results demonstrate that the developed purification and reconstitution protocol leads to the formation of a functional EAL, starting from the isolated EutB and EutC protein subunits. This result is consistent with the formation of the [(EutB–

EutC)₂]₃ EAL oligomer structure, which is evidenced by the gel electrophoresis results in Figure 5.

EPR Spectroscopy of EAL Reconstituted from Isolated EutB and EutC. The cob(II)alamin–substrate radical pair intermediate accumulates to a level of >90%, relative to active sites, in EAL during steady-state turnover on the substrate, (S)-2-aminopropanol.⁶³ The substrate radical accumulates, because the radical rearrangement is the rate-limiting step in the catalytic cycle.⁶³ The cob(II)alamin–substrate radical pair intermediate can be cryotrapped, and the characteristic radical pair line shape can be detected by using low-temperature (120 K), continuous-wave (CW) EPR spectroscopy.^{18,63,64} The line shape arises from the isotropic *J* coupling and anisotropic dipolar coupling of the two unpaired electron spins, for a defined, rigid orientation of the substrate radical and cob(II)-alamin over an 11 Å electron–electron separation,^{18,65} and is therefore exquisitely sensitive to the structure of the protein in the active site region.^{18,64} Figure 7 shows the characteristic

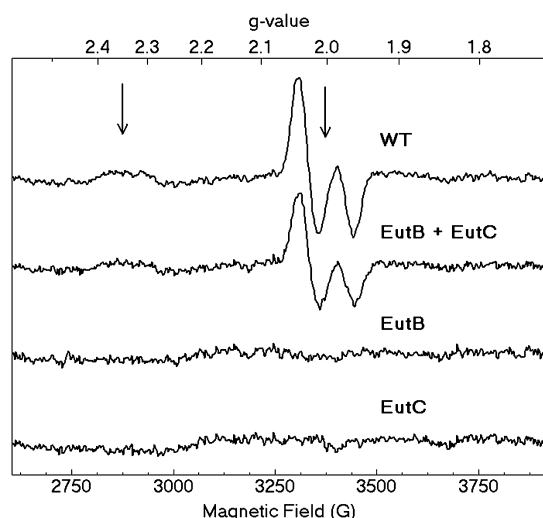


Figure 7. X-Band CW EPR spectra of the cryotrapped cob(II)alamin–substrate radical pair in EAL in different protein samples. Spectra are as follows: WT, EALH6 protein; EutB + EutC, EAL enzyme reconstituted from purified EutB and EutC subunits; EutB, purified EutB subunit, alone; EutC, purified EutC subunit, alone. Arrows show $g = 2.35$ [g_{\perp} for cob(II)alamin] and $g = 2.0023$ (free electron g value) features. Experimental conditions: microwave frequency, 9.458 GHz; temperature, 120 K; microwave power, 2.04 mW; magnetic field modulation, 10.0 G; modulation frequency, 100 kHz; scan rate, 71 G/s; time constant, 10.24 ms.

native cob(II)alamin–substrate radical pair line shape, with a $g_{\perp} = 2.35$ feature for cob(II)alamin and a $g \sim 2.00$ feature for the substrate radical, as reported previously.¹⁸ Figure 7 shows that the characteristic cob(II)alamin–substrate radical pair line shape is also present in the reconstituted EAL sample and that the individual EutB and EutC proteins did not support the formation of paramagnetic species. The spectra of wild-type EAL and reconstituted EAL overlap, upon rescaling the wild-type spectrum by a factor of 0.73. The results suggest the following. (a) The functional reconstitution of EAL from isolated EutB and EutC subunits is achieved at a level of 73%, relative to the EALH6 enzyme. This value is consistent with a reduction in k_{cat} of 70% (Table 1). The origin of the 70% level of functional reconstitution may be the partial aggregation of EutC, as evidenced on the nondenaturing PAGE gel (Figure 5,

lane 4). (b) The appearance of the radical pair indicates that the free energy surface for the reaction, and specifically, rate limitation by the radical rearrangement step, is the same in the reconstituted and control EAL oligomers. (c) The structure of EAL in the active site region of the reconstituted enzyme is comparable to the structure in native EAL.

CONCLUSIONS

A homology model for the *S. typhimurium* EAL is constructed from the X-ray crystallographic structure of EAL from *E. coli*²⁷ and demonstrated to be robust by statistical criteria. The model structure is used to describe the hierarchy of EutB and EutC subunit interactions that construct the native EAL oligomer and, specifically, to address the long-standing^{29,31} challenge of reconstitution of the functional oligomer from isolated, purified subunits. The model predicts that (EutB)₂ oligomer assembly will occur from isolated EutB and that this hexameric structure will template the formation of the complete, native [(EutB–EutC)₂]₃ oligomer. SDS–PAGE and nondenaturing PAGE and gel filtration verify this prediction. The homology model also shows multiple cysteine residues in the subunit–subunit contact surfaces of EutB and EutC, which predicts that formation of the non-native, cystine disulfide bond interferes with reconstitution. This is verified by the activation of steady-state enzyme activity, by addition of disulfide reducing agents during purification of the isolated EutB and EutC, and during reconstitution of the EAL oligomer. The significant static charge interactions at the interface between EutB (net negative) and EutC (net positive) in the heterodimer interface dictate low-ionic strength conditions during reconstitution. The angstrom-scale congruence of the reconstituted and native EAL in the active site region, which is revealed by EPR spectroscopy, demonstrates correct assembly of the catalytic machinery. Overall, the hierarchy of subunit interactions and microscopic features of the contact surfaces, which are revealed by the homology model, guides the establishment of a refined genetic and biochemical approach to reconstitution of the functional [(EutB–EutC)₂]₃ EAL oligomer and provides a rationale for the structure. This is a prerequisite for further advances in understanding the molecular mechanism of EAL catalysis, by using site-directed mutagenesis, and for insight into approaches to therapeutic manipulation of eut-associated metabolosome and disease pathways.

ASSOCIATED CONTENT

Supporting Information

Figures S1–S10 and Tables S1 and S2. This material is available free of charge via the Internet at <http://pubs.acs.org>.

AUTHOR INFORMATION

Corresponding Author

*Department of Physics, N201 Mathematics and Science Center, 400 Dowman Dr., Emory University, Atlanta, GA 30322-2430. E-mail: kwarncke@physics.emory.edu. Phone: (404) 727-2975. Fax: (404) 727-0873.

Author Contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

Funding

Research reported in this publication was supported by the National Institute of Diabetes and Digestive and Kidney

Diseases of the National Institutes of Health under Award Number R01 DK054514. The content is solely the responsibility of the authors and does not necessarily represent the official views of the National Institutes of Health. The purchase of the Bruker E500 EPR spectrometer was funded by the National Center for Research Resources of the National Institutes of Health under Award Number RR17767, and by Emory University.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We are grateful to Dr. Kim Gernert [Director of the Biomolecular Computing Resource at Emory University School of Medicine (BimCore)] for protein modeling advice and assistance.

ABBREVIATIONS

AdoCbl, 5'-deoxyadenosylcobalamin; β -ME, β -mercaptoethanol; CW, continuous-wave; DTT, dithiothreitol; EAL, ethanolamine ammonia-lyase; eut, ethanolamine utilization; EPR, electron paramagnetic resonance; IPTG, isopropyl β -D-1-thiogalactopyranoside; KP_i, potassium phosphate; PDB, Protein Data Bank; TAE, tris-acetate-EDTA; TCEP, tris(2-carboxyethyl)phosphine.

REFERENCES

- (1) Bandarian, V., and Reed, G. H. (1999) Ethanolamine ammonia-lyase. In *Chemistry and Biochemistry of B12* (Banerjee, R., Ed.) Wiley, New York.
- (2) Frey, P. A. (2010) Cobalamin coenzymes in enzymology. In *Comprehensive Natural Products II. Chemistry and Biology* (Mander, L., and Lui, H.-W., Eds.) pp 501–546, Elsevier, Oxford, U.K.
- (3) Hubbard, B. K., Gulick, A. M., Babbitt, P. C., Rayment, I., and Gerlt, J. A. (1999) Evolution of enzymatic activities in the enolase superfamily: Mechanism, structure, and metabolic context of glucarate dehydratase from *Escherichia coli*. *FASEB J.* 13, A1446.
- (4) Banerjee, R. (1999) *Chemistry and Biochemistry of B12*, Wiley, New York.
- (5) Brown, K. L. (2005) Chemistry and Enzymology of Vitamin B12. *Chem. Rev.* 105, 2075–2149.
- (6) Toraya, T. (2003) Radical Catalysis in Coenzyme B12-Dependent Isomerization (Eliminating) Reactions. *Chem. Rev.* 103, 2095–2127.
- (7) Bradbeer, C. (1965) Clostridial Fermentations of Choline and Ethanolamine. I. Preparation and Properties of Cell-Free Extracts. *J. Biol. Chem.* 240, 4669–4674.
- (8) Kofoid, E., Rappleye, C., Stojilkovic, I., and Roth, J. (1999) The 17-gene ethanolamine (eut) operon of *Salmonella typhimurium* encodes five homologues of carboxysome shell proteins. *J. Bacteriol.* 181, 5317–5329.
- (9) Brinsmade, S. R., Paldon, T., and Escalante-Semerena, J. C. (2005) Minimal functions and physiological conditions required for growth of *Salmonella enterica* on ethanolamine in the absence of the metabolosome. *J. Bacteriol.* 187, 8039–8046.
- (10) Roof, D. M., and Roth, J. R. (1988) Ethanolamine Utilization in *Salmonella typhimurium*. *J. Bacteriol.* 170, 3855–3863.
- (11) Roof, D. M., and Roth, J. R. (1989) Functions Required for Vitamin-B12-Dependent Ethanolamine Utilization in *Salmonella typhimurium*. *J. Bacteriol.* 171, 3316–3323.
- (12) Penrod, J. T., and Roth, J. R. (2006) Conserving a volatile metabolite: A role for carboxysome-like organelles in *Salmonella enterica*. *J. Bacteriol.* 188, 2865–2874.
- (13) Garsin, D. A. (2010) Ethanolamine utilization in bacterial pathogens: Roles and regulation. *Nat. Rev. Microbiol.* 8, 290–295.

- (14) Kendall, M. M., Gruber, C. C., Parker, C. T., and Sperandio, V. (2012) Ethanolamine controls expression of genes encoding components involved in interkingdom signaling and virulence in enterohemorrhagic *Escherichia coli* O157:H7. *mBio*, 3.
- (15) Srikumar, S., and Fuchs, T. M. (2011) Ethanolamine utilization contributes to proliferation of *Salmonella enterica* serovar *Typhimurium* in food and in nematodes. *Appl. Environ. Microbiol.* 77, 281–290.
- (16) Li, H., Kristensen, D. M., Coleman, M. K., and Mushegian, A. (2009) Detection of biochemical pathways by probabilistic matching of phyletic vectors. *PLoS One* 4, e5326.
- (17) Stubbe, J., and van der Donk, W. (1998) Protein radicals in enzyme catalysis. *Chem. Rev.* 98, 705–762.
- (18) Canfield, J. M., and Warncke, K. (2002) Geometry of reactant centers in the Co-II-substrate radical pair state of coenzyme B12-dependent ethanolamine deaminase determined by using orientation-selection-ESEEM spectroscopy. *J. Phys. Chem. B* 106, 8831–8841.
- (19) Canfield, J. M., and Warncke, K. (2005) Active site reactant center geometry in the Co(II)-product radical pair state of coenzyme B12-dependent ethanolamine deaminase determined by using orientation-selection electron spin-echo envelope modulation spectroscopy. *J. Phys. Chem. B* 109, 3053–3064.
- (20) LoBrutto, R., Bandarian, V., Magnusson, O. T., Chen, X., Schramm, V. L., and Reed, G. H. (2001) 5'-Deoxyadenosine contacts the substrate radical intermediate in the active site of ethanolamine ammonia-lyase: ²H and ¹³C electron nuclear double resonance studies. *Biochemistry* 40, 9–14.
- (21) Warncke, K., and Utada, A. S. (2001) Interaction of the substrate radical and the 5'-deoxyadenosine-5'-methyl group in vitamin B12 coenzyme-dependent ethanolamine deaminase. *J. Am. Chem. Soc.* 123, 8564–8572.
- (22) Bender, G., Poyner, R. R., and Reed, G. H. (2008) Identification of the Substrate Radical Intermediate Derived from Ethanolamine during Catalysis by Ethanolamine Ammonia-Lyase. *Biochemistry* 47, 11360–11366.
- (23) Poyner, R. R., Anderson, M. A., Bandarian, V., Cleland, W. W., and Reed, G. H. (2006) Probing nitrogen-sensitive steps in the free-radical-mediated deamination of amino alcohols by ethanolamine ammonia-lyase. *J. Am. Chem. Soc.* 128, 7120–7121.
- (24) Wang, M., and Warncke, K. (2008) Kinetic and thermodynamic characterization of Co-II-substrate radical pair formation in coenzyme B12-dependent ethanolamine ammonia-lyase in a cryosolvent system by using time-resolved, full-spectrum continuous-wave electron paramagnetic resonance spectroscopy. *J. Am. Chem. Soc.* 130, 4846–4858.
- (25) Zhu, C., and Warncke, K. (2008) Reaction of the Co(II)-Substrate Radical Pair Catalytic Intermediate in Coenzyme B12-Dependent Ethanolamine Ammonia-Lyase in Frozen Aqueous Solution from 190 to 217 K. *Biophys. J.* 95, 5890–5900.
- (26) Zhu, C., and Warncke, K. (2010) Kinetic Isolation and Characterization of the Radical Rearrangement Step in Coenzyme B12-Dependent Ethanolamine Ammonia-Lyase. *J. Am. Chem. Soc.* 132, 9610–9615.
- (27) Shibata, N., Tamagaki, H., Hieda, N., Akita, K., Komori, H., Shomura, Y., Terawaki, S., Mori, K., Yasuoka, N., Higuchi, Y., and Toraya, T. (2010) Crystal Structures of Ethanolamine Ammonia-lyase Complexed with Coenzyme B12 Analogs and Substrates. *J. Biol. Chem.* 285, 26484–26493.
- (28) Sun, L., and Warncke, K. (2006) Comparative model of EutB from coenzyme B12-dependent ethanolamine ammonia-lyase reveals a $\beta_8\alpha_8$, TIM-barrel fold and radical catalytic site structural features. *Proteins* 64, 308–319.
- (29) Faust, L. P., and Babior, B. M. (1992) Overexpression, Purification, and Some Properties of the AdoCbl-Dependent Ethanolamine Ammonia-Lyase from *Salmonella typhimurium*. *Arch. Biochem. Biophys.* 294, 50–54.
- (30) Gerlt, J. A., and Raushel, F. M. (2003) Evolution of function in ($\beta\alpha$)₈-barrel enzymes. *Curr. Opin. Chem. Biol.* 7, 252–264.
- (31) Faust, L. R. P., Connor, J. A., Roof, D. M., Hoch, J. A., and Babior, B. M. (1990) Cloning, Sequencing, and Expression of the

Genes Encoding the Adenosylcobalamin-Dependent Ethanolamine Ammonia-Lyase of *Salmonella typhimurium*. *J. Biol. Chem.* 265, 12462–12466.

(32) Bandarian, V., and Reed, G. H. (1999) Hydrazine cation radical in the active site of ethanolamine ammonia-lyase: Mechanism-based inactivation by hydroxyethylhydrazine. *Biochemistry* 38, 12394–12402.

(33) Guruprasad, K., Reddy, B. V. B., and Pandit, M. W. (1990) Correlation between Stability of a Protein and Its Dipeptide Composition: A Novel-Approach for Predicting In vivo Stability of a Protein from Its Primary Sequence. *Protein Eng.* 4, 155–161.

(34) Akita, K., Hieda, N., Baba, N., Kawaguchi, S., Sakamoto, H., Nakanishi, Y., Yamanishi, M., Mori, K., and Toraya, T. (2010) Purification and some properties of wild-type and N-terminal-truncated ethanolamine ammonia-lyase of *Escherichia coli*. *J. Biochem.* 147, 83–93.

(35) Goujon, M., McWilliam, H., Li, W. Z., Valentin, F., Squizzato, S., Paern, J., and Lopez, R. (2010) A new bioinformatics analysis tools framework at EMBL-EBI. *Nucleic Acids Res.* 38, W695–W699.

(36) Larkin, M. A., Blackshields, G., Brown, N. P., Chenna, R., McGettigan, P. A., McWilliam, H., Valentin, F., Wallace, I. M., Wilm, A., Lopez, R., Thompson, J. D., Gibson, T. J., and Higgins, D. G. (2007) Clustal W and clustal X version 2.0. *Bioinformatics* 23, 2947–2948.

(37) Eswar, N., Webb, B., Marti-Renom, M. A., Madhusudhan, M. S., Eramian, D., Shen, M. Y., Pieper, U., and Sali, A. (2007) Comparative protein structure modeling using MODELLER. *Current Protocols in Protein Science*, Chapter 2, Unit 2, 9, Wiley, New York.

(38) Shen, M. Y., and Sali, A. (2006) Statistical potential for assessment and prediction of protein structures. *Protein Sci.* 15, 2507–2524.

(39) Laskowski, R. A., Macarthur, M. W., Moss, D. S., and Thornton, J. M. (1993) Procheck: A Program to Check the Stereochemical Quality of Protein Structures. *J. Appl. Crystallogr.* 26, 283–291.

(40) Ahmad, S., Gromiha, M., Fawareh, H., and Sarai, A. (2004) ASAView: Database and tool for solvent accessibility representation in proteins. *BMC Bioinf.* 5, 51.

(41) Kabsch, W., and Sander, C. (1983) Dictionary of protein secondary structure: Pattern recognition of hydrogen-bonded and geometrical features. *Biopolymers* 22, 2577–2637.

(42) Baker, N. A., Sept, D., Joseph, S., Holst, M. J., and McCammon, J. A. (2001) Electrostatics of nanosystems: Application to microtubules and the ribosome. *Proc. Natl. Acad. Sci. U.S.A.* 98, 10037–10041.

(43) Dolinsky, T. J., Czodrowski, P., Li, H., Nielsen, J. E., Jensen, J. H., Klebe, G., and Baker, N. A. (2007) PDB2PQR: Expanding and upgrading automated preparation of biomolecular structures for molecular simulations. *Nucleic Acids Res.* 35, W522–W525.

(44) Dolinsky, T. J., Nielsen, J. E., McCammon, J. A., and Baker, N. A. (2004) PDB2PQR: An automated pipeline for the setup of Poisson-Boltzmann electrostatics calculations. *Nucleic Acids Res.* 32, W665–W667.

(45) Finn, R. D., Mistry, J., Tate, J., Coghill, P., Heger, A., Pollington, J. E., Gavin, O. L., Gunasekaran, P., Ceric, G., Forslund, K., Holm, L., Sonnhammer, E. L., Eddy, S. R., and Bateman, A. (2010) The Pfam protein families database. *Nucleic Acids Res.* 38, D211–D222.

(46) Bradford, M. M. (1976) Rapid and sensitive method for the quantitation of microgram quantities of protein utilizing the principle of protein-dye binding. *Anal. Biochem.* 72, 248–254.

(47) Gasteiger, E., Hoogland, C., Gattiker, A., Duvaud, S., Wilkins, M. R., Appel, R. D., and Bairoch, A. (2005) Protein Identification and Analysis Tools on the ExPASy Server. In *The Proteomics Protocols Handbook* (Walker, J. M., Ed.) pp 571–607, Humana Press, Totowa, NJ.

(48) Laemmli, U. K. (1970) Cleavage of Structural Proteins during Assembly of Head of Bacteriophage-T4. *Nature* 227, 680–685.

(49) Ornstein, L. (1964) Disc Electrophoresis. I. Background and Theory. *Ann. N.Y. Acad. Sci.* 121, 321–349.

(50) Kaplan, B. H., and Stadtman, E. R. (1968) Ethanolamine Deaminase a Cobamide Coenzyme-Dependent Enzyme. I. Purification Assay and Properties of Enzyme. *J. Biol. Chem.* 243, 1787.

(51) Sali, A., and Blundell, T. L. (1993) Comparative protein modelling by satisfaction of spatial restraints. *J. Mol. Biol.* 234, 779–815.

(52) Engh, R. A., and Huber, R. (1991) Accurate Bond and Angle Parameters for X-ray Protein-Structure Refinement. *Acta Crystallogr. A* 47, 392–400.

(53) Mitov, M. I., Greaser, M. L., and Campbell, K. S. (2009) GelBandFitter: A computer program for analysis of closely spaced electrophoretic and immunoblotted bands. *Electrophoresis* 30, 848–851.

(54) Samuelson, J. C. (2011) Recent developments in difficult protein expression: A guide to *E. coli* strains, promoters, and relevant host mutations. *Methods Mol. Biol.* 705, 195–209.

(55) Arakawa, T., and Kita, Y. (2000) Protection of bovine serum albumin from aggregation by Tween 80. *J. Pharm. Sci.* 89, 646–651.

(56) Choudhary, S., Quin, M. B., Sanders, M. A., Johnson, E. T., and Schmidt-Dannert, C. (2012) Engineered protein nano-compartments for targeted enzyme localization. *PLoS One* 7, e33342.

(57) Derman, A. I., Prinz, W. A., Belin, D., and Beckwith, J. (1993) Mutations that allow disulfide bond formation in the cytoplasm of *Escherichia coli*. *Science* 262, 1744–1747.

(58) Cleland, W. W. (1964) Dithiothreitol New Protective Reagent for SH Groups. *Biochemistry* 3, 480–482.

(59) Jocelyn, P. C. (1987) Chemical-Reduction of Disulfides. *Methods Enzymol.* 143, 246–256.

(60) Burns, J. A., Butler, J. C., Moran, J., and Whitesides, G. M. (1991) Selective Reduction of Disulfides by Tris(2-Carboxyethyl)-Phosphine. *J. Org. Chem.* 56, 2648–2650.

(61) Getz, E. B., Xiao, M., Chakrabarty, T., Cooke, R., and Selvin, P. R. (1999) A comparison between the sulfhydryl reductants tris(2-carboxyethyl)phosphine and dithiothreitol for use in protein biochemistry. *Anal. Biochem.* 273, 73–80.

(62) Fersht, A. (1999) *Structure and mechanism in protein science*, W. H. Freeman and Co., New York.

(63) Babor, B. M., Moss, T. H., Orme-Johnson, W. H., and Beinert, H. (1974) Mechanism of Action of Ethanolamine Ammonia-Lyase, a B12-Dependent Enzyme: Participation of Paramagnetic Species in Catalytic Deamination of 2-Aminopropanol. *J. Biol. Chem.* 249, 4537–4544.

(64) Gerfen, G. J. (1999) EPR spectroscopy of B12-dependent enzymes. In *Chemistry and Biochemistry of B12* (Banerjee, R., Ed.) pp 165–195, John Wiley and Sons, New York.

(65) Pilbrow, J. (1982) EPR of B12-Dependent Enzyme Reactions and Related Systems. In *B12* (Dolphin, D., Ed.) pp 431–462, Wiley, New York.